

Environmental Protection Agency

§ 1065.640

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[73 FR 37324, June 30, 2008, as amended at 73 FR 59330, Oct. 8, 2008; 75 FR 23045, Apr. 30, 2010; 76 FR 57453, Sept. 15, 2011]

EFFECTIVE DATE NOTE: At 78 FR 36398, June 17, 2013, §1065.610 was amended by revising paragraph (c)(3), effective Aug. 16, 2013. For the convenience of the user, the revised text is set forth as follows:

§ 1065.610 Duty cycle generation.

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(c) * * *

(3) *Intermediate speed.* If your normalized duty cycle specifies a speed as “intermediate speed,” use your torque-versus-speed curve to determine the speed at which maximum torque occurs. This is peak torque speed. If maximum torque occurs in a flat region of the torque-versus-speed curve, your peak torque speed is the midpoint between the lowest and highest speeds at which the trace reaches the flat region. For purposes of this paragraph (c)(3), a flat region is one in which measured torque values are within 2% of the maximum recorded value. Identify your reference intermediate speed as one of the following values:

- (i) Peak torque speed if it is between (60 and 75) % of maximum test speed.
- (ii) 60% of maximum test speed if peak torque speed is less than 60% of maximum test speed.
- (iii) 75% of maximum test speed if peak torque speed is greater than 75% of maximum test speed.

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§ 1065.630 1980 international gravity formula.

The acceleration of Earth’s gravity, a_g , varies depending on your location. Calculate a_g at your latitude, as follows:

$$a_g = 9.7803267715 \cdot [1 + s \\ 5.2790414 \cdot 10^{-3} \cdot \sin^2(\theta) + \\ 2.32718 \cdot 10^{-5} \cdot \sin^4(\theta) + \\ 1.262 \cdot 10^{-7} \cdot \sin^6(\theta) + \\ 7 \cdot 10^{-10} \cdot \sin^8(\theta)] \quad \text{Eq. 1065.630-1}$$

Where:

θ = Degrees north or south latitude.

Example:

$\theta = 45^\circ$

$$a_g = 9.7803267715 \cdot (1 + \\ 5.2790414 \cdot 10^{-3} \cdot \sin^2(45) + \\ 2.32718 \cdot 10^{-5} \cdot \sin^4(45) + \\ 1.262 \cdot 10^{-7} \cdot \sin^6(45) + \\ 7 \cdot 10^{-10} \cdot \sin^8(45)) \\ a_g = 9.8178291229 \text{ m/s}^2$$

§ 1065.640 Flow meter calibration calculations.

This section describes the calculations for calibrating various flow meters. After you calibrate a flow meter using these calculations, use the calculations described in §1065.642 to calculate flow during an emission test. Paragraph (a) of this section first describes how to convert reference flow meter outputs for use in the calibration equations, which are presented on a molar basis. The remaining paragraphs describe the calibration calculations that are specific to certain types of flow meters.

(a) *Reference meter conversions.* The calibration equations in this section use molar flow rate, \dot{n}_{ref} , as a reference quantity. If your reference meter outputs a flow rate in a different quantity, such as standard volume rate, \dot{V}_{stdref} , actual volume rate, \dot{V}_{actref} , or mass rate, \dot{m}_{ref} , convert your reference meter output to a molar flow rate using the following equations, noting that while values for volume rate, mass rate, pressure, temperature, and molar mass may change during an emission test, you should ensure that they are as constant as practical for each individual set point during a flow meter calibration:

$$\dot{n}_{\text{ref}} = \frac{\dot{V}_{\text{stdref}} \cdot P_{\text{std}}}{T_{\text{std}} \cdot R} = \frac{\dot{V}_{\text{actref}} \cdot P_{\text{act}}}{T_{\text{act}} \cdot R} = \frac{\dot{m}_{\text{ref}}}{M_{\text{mix}}}$$

Eq. 1065.640-1

Where:

 \dot{n}_{ref} = reference molar flow rate. \dot{V}_{stdref} = reference volume flow rate, corrected to a standard pressure and a standard temperature. \dot{V}_{actref} = reference volume flow rate at the actual pressure and temperature of the flow rate. \dot{m}_{ref} = reference mass flow. P_{std} = standard pressure. P_{act} = actual pressure of the flow rate. T_{std} = standard temperature. T_{act} = actual temperature of the flow rate. R = molar gas constant. M_{mix} = molar mass of the flow rate.

Example 1:

 $\dot{V}_{\text{stdref}} = 1000.00 \text{ ft}^3/\text{min} = 0.471948 \text{ m}^3/\text{s}$ $P = 29.9213 \text{ in Hg @ } 32^\circ\text{F} = 101325 \text{ Pa}$ $T = 68.0^\circ\text{F} = 293.15 \text{ K}$ $R = 8.314472 \text{ J}/(\text{mol} \cdot \text{K})$

$$\dot{n}_{\text{ref}} = \frac{0.471948 \cdot 101325}{293.15 \cdot 8.314472}$$

 $\dot{n}_{\text{ref}} = 19.619 \text{ mol/s}$ $\dot{M}_{\text{mix}} = 28.7805 \text{ g/mol}$

Example 2:

 $\dot{m}_{\text{ref}} = 17.2683 \text{ kg/min} = 287.805 \text{ g/s}$

$$\dot{n}_{\text{ref}} = \frac{287.805}{28.7805}$$

 $\dot{n}_{\text{ref}} = 10.0000 \text{ mol/s}$ (b) *PDP calibration calculations.* For each restrictor position, calculate the

following values from the mean values determined in § 1065.340, as follows:

(1) PDP volume pumped per revolution, V_{rev} (m^3/r):

$$V_{\text{rev}} = \frac{\bar{\dot{n}}_{\text{ref}} \cdot R \cdot \bar{T}_{\text{in}}}{\bar{P}_{\text{in}} \cdot \bar{f}_{\text{nPDP}}}$$

Eq. 1065.640-2

Example:

 $\bar{\dot{n}}_{\text{ref}} = 25.096 \text{ mol/s}$ $R = 8.314472 \text{ J}/(\text{mol} \cdot \text{K})$ $\bar{T}_{\text{in}} = 299.5 \text{ K}$ $\bar{P}_{\text{in}} = 98290 \text{ Pa}$ $\bar{f}_{\text{nPDP}} = 1205.1 \text{ r/min} = 20.085 \text{ r/s}$ $V_{\text{rev}} = 0.03166 \text{ m}^3/\text{r}$ (2) PDP slip correction factor, K_s (s/r):

$$V_{rev} = \frac{25.096 \cdot 8.314472 \cdot 299.5}{98290 \cdot 20.085}$$

$$K_s = \frac{1}{\bar{f}_{nPDP}} \cdot \sqrt{\frac{\bar{P}_{out} - \bar{P}_{in}}{\bar{P}_{out}}}$$

Eq. 1065.640-3

Example: $\bar{f}_{nPDP} = 1205.1 \text{ r/min} = 20.085 \text{ r/s}$ $\bar{P}_{out} = 100.103 \text{ kPa}$ $\bar{P}_{in} = 98.290 \text{ kPa}$

$$K_s = \frac{1}{20.085} \cdot \sqrt{\frac{100.103 - 98.290}{100.103}}$$

 $K_s = 0.006700 \text{ s/r}$

(3) Perform a least-squares regression of PDP volume pumped per revolution, V_{rev} , versus PDP slip correction factor, K_s , by calculating slope, a_1 , and intercept, a_0 , as described in § 1065.602.

(4) Repeat the procedure in paragraphs (b)(1) through (3) of this section for every speed that you run your PDP.

(5) The following example illustrates these calculations:

TABLE 1 OF § 1065.640—EXAMPLE OF PDP CALIBRATION DATA

\bar{f}_{nPDP} (r/min)	a_1 (m ³ /min)	a_0 (m ³ /r)
755.0	50.43	0.056
987.6	49.86	−0.013
1254.5	48.54	0.028
1401.3	47.30	−0.061

(6) For each speed at which you operate the PDP, use the corresponding slope, a_1 , and intercept, a_0 , to calculate flow rate during emission testing as described in § 1065.642.

(c) *Venturi governing equations and permissible assumptions.* This section describes the governing equations and permissible assumptions for calibrating a venturi and calculating flow using a venturi. Because a subsonic venturi (SSV) and a critical-flow venturi (CFV) both operate similarly, their governing equations are nearly the same, except for the equation de-

scribing their pressure ratio, r (*i.e.*, r_{SSV} versus r_{CFV}). These governing equations assume one-dimensional isentropic inviscid compressible flow of an ideal gas. In paragraph (c)(4) of this section, we describe other assumptions that you may make, depending upon how you conduct your emission tests. If we do not allow you to assume that the measured flow is an ideal gas, the governing equations include a first-order correction for the behavior of a real gas; namely, the compressibility factor, Z . If good engineering judgment dictates using a value other than $Z=1$, you may either use an appropriate equation of state to determine values of Z as a function of measured pressures and temperatures, or you may develop your own calibration equations based on good engineering judgment. Note that the equation for the flow coefficient, C_f , is based on the ideal gas assumption that the isentropic exponent, γ , is equal to the ratio of specific heats, C_p/C_v . If good engineering judgment dictates using a real gas isentropic exponent, you may either use an appropriate equation of state to determine values of γ as a function of measured pressures and temperatures, or you may develop your own calibration equations based on good engineering judgment. Calculate molar flow rate, \dot{n} , as follows:

$$\dot{n} = C_d \cdot C_f \cdot \frac{A_t \cdot p_{in}}{\sqrt{Z \cdot M_{mix} \cdot R \cdot T_{in}}} \quad \text{Eq. 1065.640-4}$$

Where:

C_d = Discharge coefficient, as determined in paragraph (c)(1) of this section.

C_f = Flow coefficient, as determined in paragraph (c)(2) of this section.

A_t = Venturi throat cross-sectional area.

p_{in} = Venturi inlet absolute static pressure.

Z = Compressibility factor.

M_{mix} = Molar mass of gas mixture.

R = Molar gas constant.

T_{in} = Venturi inlet absolute temperature.

(1) Using the data collected in § 1065.340, calculate C_d using the following equation:

$$C_d = \dot{n}_{ref} \cdot \frac{\sqrt{Z \cdot M_{mix} \cdot R \cdot T_{in}}}{C_f \cdot A_t \cdot p_{in}} \quad \text{Eq. 1065.640-5}$$

Where:

\dot{n}_{ref} = A reference molar flow rate.

(2) Determine C_f using one of the following methods:

(i) For CFV flow meters only, determine C_{fCFV} from the following table based on your values for β and γ , using linear interpolation to find intermediate values:

TABLE 2 OF § 1065.640— C_{fCFV} VERSUS β AND γ
FOR CFV FLOW METERS

C_{fCFV}		
β	$\gamma_{exh} = 1.385$	$\gamma_{dexh} = \gamma_{air} = 1.399$
0.000	0.6822	0.6846
0.400	0.6857	0.6881
0.500	0.6910	0.6934
0.550	0.6953	0.6977
0.600	0.7011	0.7036
0.625	0.7047	0.7072
0.650	0.7089	0.7114

TABLE 2 OF § 1065.640— C_{fCFV} VERSUS β AND γ
FOR CFV FLOW METERS—Continued

C_{fCFV}		
β	$\gamma_{exh} = 1.385$	$\gamma_{dexh} = \gamma_{air} = 1.399$
0.675	0.7137	0.7163
0.700	0.7193	0.7219
0.720	0.7245	0.7271
0.740	0.7303	0.7329
0.760	0.7368	0.7395
0.770	0.7404	0.7431
0.780	0.7442	0.7470
0.790	0.7483	0.7511
0.800	0.7527	0.7555
0.810	0.7573	0.7602
0.820	0.7624	0.7652
0.830	0.7677	0.7707
0.840	0.7735	0.7765
0.850	0.7798	0.7828

(ii) For any CFV or SSV flow meter, you may use the following equation to calculate C_f :

$$C_f = \left[\frac{2 \cdot \gamma \cdot \left(r^{\frac{\gamma-1}{\gamma}} - 1 \right)}{(\gamma-1) \cdot \left(\beta^4 - r^{\frac{-2}{\gamma}} \right)} \right]^{\frac{1}{2}} \quad \text{Eq. 1065.640-6}$$

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Where:

γ = isentropic exponent. For an ideal gas, this is the ratio of specific heats of the gas mixture, C_p/C_v .

r = Pressure ratio, as determined in paragraph (c)(3) of this section.

β = Ratio of venturi throat to inlet diameters.

(3) Calculate r as follows:

(i) For SSV systems only, calculate r_{SSV} using the following equation:

$$r_{SSV} = 1 - \frac{\Delta p_{SSV}}{p_{in}} \quad \text{Eq. 1065.640-7}$$

Where:

Δp_{SSV} = Differential static pressure; venturi inlet minus venturi throat.

(ii) For CFV systems only, calculate r_{CFV} iteratively using the following equation:

$$r_{CFV}^{\frac{1-\gamma}{\gamma}} + \left(\frac{\gamma-1}{2} \right) \cdot \beta^4 \cdot r_{CFV}^{\frac{2}{\gamma}} = \frac{\gamma+1}{2} \quad \text{Eq. 1065.640-8}$$

(4) You may make any of the following simplifying assumptions of the governing equations, or you may use good engineering judgment to develop more appropriate values for your testing:

(i) For emission testing over the full ranges of raw exhaust, diluted exhaust and dilution air, you may assume that the gas mixture behaves as an ideal gas: $Z = 1$.

(ii) For the full range of raw exhaust you may assume a constant ratio of specific heats of $\gamma = 1.385$.

(iii) For the full range of diluted exhaust and air (e.g., calibration air or dilution air), you may assume a constant ratio of specific heats of $\gamma = 1.399$.

(iv) For the full range of diluted exhaust and air, you may assume the molar mass of the mixture is a function only of the amount of water in the dilution air or calibration air, x_{H_2O} , determined as described in §1065.645, as follows:

$$M_{mix} = M_{air} \cdot (1 - x_{H_2O}) + M_{H_2O} \cdot x_{H_2O} \quad \text{Eq. 1065.640-9}$$

Example:

$M_{air} = 28.96559 \text{ g/mol}$

$x_{H_2O} = 0.0169 \text{ mol/mol}$

$M_{H_2O} = 18.01528 \text{ g/mol}$

$M_{mix} = 28.96559 \cdot (1 - 0.0169) + 18.01528 \cdot 0.0169$

$M_{mix} = 28.7805 \text{ g/mol}$

(v) For the full range of diluted exhaust and air, you may assume a constant molar mass of the mixture, M_{mix} , for all calibration and all testing as long as your assumed molar mass differs no more than $\pm 1\%$ from the esti-

mated minimum and maximum molar mass during calibration and testing. You may assume this, using good engineering judgment, if you sufficiently control the amount of water in calibration air and in dilution air or if you remove sufficient water from both calibration air and dilution air. The following table gives examples of permissible ranges of dilution air dewpoint versus calibration air dewpoint:

TABLE 3 OF § 1065.640—EXAMPLES OF DILUTION AIR AND CALIBRATION AIR DEWPOINTS AT WHICH YOU MAY ASSUME A CONSTANT M_{mix} .

If calibration T_{dew} (°C) is...	assume the following constant M_{mix} (g/mol)...	for the following ranges of T_{dew} (°C) during emission tests ^a
dry	28.96559	dry to 18.
0	28.89263	dry to 21.
5	28.86148	dry to 22.
10	28.81911	dry to 24.
15	28.76224	dry to 26.
20	28.68685	–8 to 28.
25	28.58806	12 to 31.
30	28.46005	23 to 34.

^a Range valid for all calibration and emission testing over the atmospheric pressure range (80.000 to 103.325) kPa.

(5) The following example illustrates the use of the governing equations to calculate the discharge coefficient, C_d , of an SSV flow meter at one reference flow meter value. Note that calculating C_d for a CFV flow meter would be similar, except that C_f would be determined from Table 2 of this section or calculated iteratively using values of β and γ as described in paragraph (c)(2) of this section.

Example:

$$\dot{n}_{\text{ref}} = 57.625 \text{ mol/s}$$

$$Z = 1$$

$$M_{\text{mix}} = 28.7805 \text{ g/mol} = 0.0287805 \text{ kg/mol}$$

$$R = 8.314472 \text{ J/(mol}\cdot\text{K)}$$

$$T_{\text{in}} = 298.15 \text{ K}$$

$$A_t = 0.01824 \text{ m}^2$$

$$p_{\text{in}} = 99132.0 \text{ Pa}$$

$$\gamma = 1.399$$

$$\beta = 0.8$$

$$\Delta p = 2.312 \text{ kPa}$$

$$r_{\text{SSV}} = 1 - \frac{2.312}{99.132} = 0.977$$

$$C_f = \left[\frac{2 \cdot 1.399 \cdot \left(0.977^{\frac{1.399-1}{1.399}} - 1 \right)}{(1.399-1) \cdot \left(0.8^4 - 0.977^{\frac{-2}{1.399}} \right)} \right]^{\frac{1}{2}}$$

$$C_f = 0.274$$

$$C_d = 57.625 \cdot \frac{\sqrt{1 \cdot 0.0287805 \cdot 8.314472 \cdot 298.15}}{0.274 \cdot 0.01824 \cdot 99132.0}$$

$$C_d = 0.982$$

(d) *SSV calibration.* Perform the following steps to calibrate an SSV flow meter:

(1) Calculate the Reynolds number, $Re^{\#}$, for each reference molar flow rate, using the throat diameter of the venturi, d_t . Because the dynamic viscosity, μ , is needed to compute $Re^{\#}$, you may

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use your own fluid viscosity model to determine μ for your calibration gas (usually air), using good engineering judgment. Alternatively, you may use

the Sutherland three-coefficient viscosity model to approximate μ , as shown in the following sample calculation for $Re^\#$:

$$Re^\# = \frac{4 \cdot M_{\text{mix}} \cdot \dot{n}_{\text{ref}}}{\pi \cdot d_t \cdot \mu} \quad \text{Eq. 1065.640-10}$$

Where, using the Sutherland three-coefficient viscosity model:

$$\mu = \mu_0 \cdot \left(\frac{T_{\text{in}}}{T_0} \right)^{\frac{3}{2}} \cdot \left(\frac{T_0 + S}{T_{\text{in}} + S} \right) \quad \text{Eq. 1065.640-11}$$

Where:

μ = Dynamic viscosity of calibration gas.

μ_0 = Sutherland reference viscosity.

T_0 = Sutherland reference temperature.

S = Sutherland constant.

TABLE 4 OF § 1065.640—SUTHERLAND THREE-COEFFICIENT VISCOSITY MODEL PARAMETERS

Gas ^a	μ_0	T_0	S	Temp range within $\pm 2\%$ error	Pressure limit
	kg/(m·s)	K	K		kPa
Air	$1.716 \cdot 10^{-5}$	273	111	170 to 1,900	$\leq 1,800$
CO ₂	$1.370 \cdot 10^{-5}$	273	222	190 to 1,700	$\leq 3,600$
H ₂ O	$1.12 \cdot 10^{-5}$	350	1,064	360 to 1,500	$\leq 10,000$
O ₂	$1.919 \cdot 10^{-5}$	273	139	190 to 2,000	$\leq 2,500$
N ₂	$1.663 \cdot 10^{-5}$	273	107	100 to 1,500	$\leq 1,600$

^a Use tabulated parameters only for the pure gases, as listed. Do not combine parameters in calculations to calculate viscosities of gas mixtures.

Example:

$\mu_0 = 1.716 \cdot 10^{-5} \text{ kg/(m·s)}$

$T_0 = 273.11 \text{ K}$

$S = 110.56 \text{ K}$

$$\mu = 1.716 \cdot 10^{-5} \cdot \left(\frac{298.15}{273.11} \right)^{\frac{3}{2}} \cdot \left(\frac{273.11 + 110.56}{298.15 + 110.56} \right)$$

$\mu = 1.837 \cdot 10^{-5} \text{ kg/(m·s)}$

$M_{\text{mix}} = 28.7805 \text{ g/mol}$

$\dot{n}_{\text{ref}} = 57.625 \text{ mol/s}$

$d_t = 152.4 \text{ mm}$

$T_{\text{in}} = 298.15 \text{ K}$

$$Re^\# = \frac{4 \cdot 28.7805 \cdot 57.625}{3.14159 \cdot 152.4 \cdot 1.837 \cdot 10^{-5}}$$

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$$Re^{\#} = 7.541 \cdot 10^5$$

(2) Create an equation for C_d versus $Re^{\#}$, using paired values of $(Re^{\#}, C_d)$. For the equation, you may use any mathematical expression, including a polynomial or a power series. The following equation is an example of a commonly used mathematical expression for relating C_d and $Re^{\#}$:

$$C_d = a_0 - a_1 \cdot \sqrt{\frac{10^6}{Re^{\#}}} \quad \text{Eq. 1065.640-12}$$

(3) Perform a least-squares regression analysis to determine the best-fit coefficients to the equation and calculate the equation's regression statistics, SEE and r^2 , according to §1065.602.

(4) If the equation meets the criteria of $SEE \leq 0.5\% \cdot \dot{n}_{\text{refmax}}$ and $r^2 \geq 0.995$, you may use the equation to determine C_d for emission tests, as described in §1065.642.

(5) If the SEE and r^2 criteria are not met, you may use good engineering judgment to omit calibration data points to meet the regression statistics. You must use at least seven calibration data points to meet the criteria.

(6) If omitting points does not resolve outliers, take corrective action. For example, select another mathematical expression for the C_d versus $Re^{\#}$ equation, check for leaks, or repeat the calibration process. If you must repeat the process, we recommend applying tighter tolerances to measurements and allowing more time for flows to stabilize.

(7) Once you have an equation that meets the regression criteria, you may

use the equation only to determine flow rates that are within the range of the reference flow rates used to meet the C_d versus $Re^{\#}$ equation's regression criteria.

(e) *CFV calibration.* Some CFV flow meters consist of a single venturi and some consist of multiple venturis, where different combinations of venturis are used to meter different flow rates. For CFV flow meters that consist of multiple venturis, either calibrate each venturi independently to determine a separate discharge coefficient, C_d , for each venturi, or calibrate each combination of venturis as one venturi. In the case where you calibrate a combination of venturis, use the sum of the active venturi throat areas as A_i , the square root of the sum of the squares of the active venturi throat diameters as d_i , and the ratio of the venturi throat to inlet diameters as the ratio of the square root of the sum of the active venturi throat diameters (d_i) to the diameter of the common entrance to all of the venturis (D). To determine the C_d for a single venturi or a single combination of venturis, perform the following steps:

(1) Use the data collected at each calibration set point to calculate an individual C_d for each point using Eq. 1065.640–4.

(2) Calculate the mean and standard deviation of all the C_d values according to Eqs. 1065.602–1 and 1065.602–2.

(3) If the standard deviation of all the C_d values is less than or equal to 0.3% of the mean C_d , use the mean C_d in Eq. 1065.642–6, and use the CFV only up to the highest r measured during calibration using the following equation:

$$r = 1 - \frac{\Delta p_{\text{CFV}}}{P_{\text{in}}}$$

Eq. 1065.640-13

Where:

Δp_{CFV} = Differential static pressure; venturi inlet minus venturi outlet.

(4) If the standard deviation of all the C_d values exceeds 0.3% of the mean C_d ,

omit the C_d values corresponding to the data point collected at the highest r measured during calibration.

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(5) If the number of remaining data points is less than seven, take corrective action by checking your calibration data or repeating the calibration process. If you repeat the calibration process, we recommend checking for leaks, applying tighter tolerances to measurements and allowing more time for flows to stabilize.

(6) If the number of remaining C_d values is seven or greater, recalculate the mean and standard deviation of the remaining C_d values.

(7) If the standard deviation of the remaining C_d values is less than or equal to 0.3% of the mean of the remaining C_d , use that mean C_d in Eq 1065.642-6, and use the CFV values only up to the highest r associated with the remaining C_d .

(8) If the standard deviation of the remaining C_d still exceeds 0.3% of the mean of the remaining C_d values, re-

peat the steps in paragraph (e)(4) through (8) of this section.

[70 FR 40516, July 13, 2005, as amended at 73 FR 37326, June 30, 2008; 73 FR 59331, Oct. 8, 2008; 75 FR 23045, Apr. 30, 2010; 75 FR 68464, Nov. 8, 2010; 76 FR 57455, Sept. 15, 2011]

§ 1065.642 SSV, CFV, and PDP molar flow rate calculations.

This section describes the equations for calculating molar flow rates from various flow meters. After you calibrate a flow meter according to §1065.640, use the calculations described in this section to calculate flow during an emission test.

(a) *PDP molar flow rate.* Based upon the speed at which you operate the PDP for a test interval, select the corresponding slope, a_1 , and intercept, a_0 , as calculated in §1065.640, to calculate molar flow rate, \dot{n} as follows:

$$\dot{n} = f_{\text{nPDP}} \cdot \frac{p_{\text{in}} \cdot V_{\text{rev}}}{R \cdot T_{\text{in}}}$$

Eq. 1065.642-1

Where:

$$V_{\text{rev}} = \frac{a_1}{f_{\text{nPDP}}} \cdot \sqrt{\frac{p_{\text{out}} - p_{\text{in}}}{p_{\text{out}}}} + a_0$$

Eq. 1065.642-2

Example:

$a_1 = 50.43 \text{ (m}^3\text{/min)} = 0.8405 \text{ (m}^3\text{/s)}$
 $f_{\text{nPDP}} = 755.0 \text{ r/min} = 12.58 \text{ r/s}$
 $p_{\text{out}} = 99950 \text{ Pa}$
 $p_{\text{in}} = 98575 \text{ Pa}$

$a_0 = 0.056 \text{ (m}^3\text{/r)}$
 $R = 8.314472 \text{ J/(mol} \cdot \text{K)}$
 $T_{\text{in}} = 323.5 \text{ K}$
 $C_p = 1000 \text{ (J/m}^3\text{)/kPa}$
 $C_t = 60 \text{ s/min}$

$$V_{\text{rev}} = \frac{0.8405}{12.58} \cdot \sqrt{\frac{99950 - 98575}{99950}} + 0.056$$

$V_{\text{rev}} = 0.06383 \text{ m}^3\text{/r}$